

**AMENDMENTS TO THE CLAIMS:**

This listing of claims will replace all prior versions, and listings, of claims in the application:

**LISTING OF CLAIMS:**

- 1-3. (Cancelled)
4. (Previously Presented) A ferromagnetic perovskite oxide having the formula  $(\text{Ba}_{0.95}\text{Fe}_{0.05})\text{TiO}_3$ , wherein the oxide has a saturation magnetization of about  $0.10 \mu_B/\text{mol Fe}$  at 300K, and a coercive field of about 16 Oe at 300K.
5. (Previously Presented) A ferromagnetic perovskite oxide having the formula  $(\text{Ca}_{0.95}\text{Fe}_{0.05})\text{TiO}_3$ , wherein the oxide has a saturation magnetization of about  $0.11 \mu_B/\text{mol Fe}$  at 300K, and a coercive field of about 12 Oe at 300K.
6. (Previously Presented) A ferromagnetic perovskite oxide having the formula  $(\text{Ba}_{0.95}\text{Fe}_{0.05})\text{ZrO}_3$ , wherein the oxide has a saturation magnetization of about  $0.11 \mu_B/\text{mol Fe}$  at 300K, and a coercive field of about 25 Oe at 300K.
7. (Previously Presented) A ferromagnetic perovskite oxide having the formula  $(\text{Ca}_{0.95}\text{Fe}_{0.05})\text{ZrO}_3$ , wherein the oxide has a saturation magnetization of about  $0.12 \mu_B/\text{mol Fe}$  at 300K, and a coercive field of about 4.5 Oe at 300K.
8. (Previously Presented) A ferromagnetic perovskite oxide having the formula  $(\text{Ba}_{0.95}\text{Fe}_{0.05})\text{HfO}_3$ , wherein the oxide has a saturation magnetization of about  $0.125 \mu_B/\text{mol Fe}$  at 300K, and a coercive field of about 20 Oe at 300K.

9. (Currently Amended) ~~The material composition of claim 2 having specific formula (Ca<sub>0.95</sub>Fe<sub>0.05</sub>)HfO<sub>3</sub>, wherein said saturation magnetization about 0.12 B/mol Fe at 300K, and the coercive fields about 7 Oe at 300K. A ferromagnetic perovskite oxide having the formula (Ca<sub>0.95</sub>Fe<sub>0.05</sub>)HfO<sub>3</sub>, wherein the oxide has a saturation magnetization of about 0.12  $\mu_B$ /mol Fe at 300K, and a coercive field of about 7 Oe at 300K.~~

10. (Cancelled)

11.-14. (Cancelled)

15. (Previously Presented) A ferromagnetic perovskite oxide having the formula La(Mo<sub>0.25</sub>Fe<sub>0.75</sub>)O<sub>3</sub>, wherein the magnetic Curie temperature of the oxide is as high as 940 K, and wherein the oxide has a coercive field of about 238 Oe at 300K.

16.-18. (Cancelled Herein)

19. (Previously Presented) A ferromagnetic perovskite oxide having the formula (Ba<sub>1-x</sub>Fe<sub>x</sub>)TiO<sub>3</sub>, where x ranges from 0 to 0.15, and wherein the oxide has a saturation magnetization of about 0.10  $\mu_B$ /mol Fe at 300K, and a coercive field of about 16 Oe at 300K.

20. (Previously Presented) A ferromagnetic perovskite oxide having the formula (Ca<sub>1-x</sub>Fe<sub>x</sub>)TiO<sub>3</sub>, where x ranges from 0 to 0.15, and wherein the oxide has a saturation magnetization of about 0.11  $\mu_B$ /mol Fe at 300K, and a coercive field of about 12 Oe at 300K.

21. (Previously Presented) A ferromagnetic perovskite oxide having the formula (Ba<sub>1-x</sub>Fe<sub>x</sub>)ZrO<sub>3</sub>, where x ranges from 0 to 0.15, and wherein the oxide has a saturation magnetization of about 0.11  $\mu_B$ /mol Fe at 300K, and a coercive field of about 25 Oe at 300K.

22. (Previously Presented) A ferromagnetic perovskite oxide having the formula  $(\text{Ca}_{1-x}\text{Fe}_x)\text{ZrO}_3$ , where  $x$  ranges from 0 to 0.15, and wherein the oxide has a saturation magnetization of about  $0.12 \mu_B/\text{mol Fe}$  at 300K, and a coercive field of about 4.5 Oe at 300K.

23. (Previously Presented) A ferromagnetic perovskite oxide having the formula  $(\text{Ba}_{1-x}\text{Fe}_x)\text{HfO}_3$ , where  $x$  ranges from 0 to 0.15, and wherein the oxide has a saturation magnetization of about  $0.125 \mu_B/\text{mol Fe}$  at 300K, and a coercive field of about 20 Oe at 300K.

24. (Previously Presented) A ferromagnetic perovskite oxide having the formula  $(\text{Ca}_{1-x}\text{Fe}_x)\text{HfO}_3$ , where  $x$  ranges from 0 to 0.15, and wherein the oxide has a saturation magnetization of about  $0.12 \mu_B/\text{mol Fe}$  at 300K, and a coercive field of about 7 Oe at 300K.